

Research of Mosses Accumulation Properties Used for Assessment of Regional and Local Atmospheric Pollution

Nadezhda K. Ryzhakova¹, Natalya S. Rogova¹, Alex L. Borisenko²

¹Institute of Physics and Technology, Tomsk Polytechnic University, Tomsk, RF ²Institute of Biology, Tomsk State University, Tomsk, RF

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The monitoring of atmospheric heavy metal and other toxic element depositions by using widespread bryophytes as biomonitors has been widely used. Choosing most appropriate moss species in relation to their accumulation properties is the main goal of this research. The accumulation of heavy metals and other toxic elements by widespread mosses of midland have been studied. The research is focused on assessing accumulation properties of 4 species of terrestrial moss, 4 species of paludal moss and 2 species of epiphytic moss. The concentrations of 32 elements have been determined in moss samples by neutron activation analysis (NAA) and atom emission spectrometry (AES). Interspecies and intraspecies comparison revealed significant differences in accumulation properties. The accumulation capacity of Dicranum polysetum was higher than other terrestrial mosses and Aulacomnium palustre had higher accumulation capacity than other paludal mosses. These moss species have been used for monitoring atmospheric pollutants in an immense territory, particularly for research of transboundary transfer of heavy metal pollution. The accumulation property of epiphytic moss was higher than others. The epiphytic moss could be found on the bark of old trees (aspens, poplars, birch) that are of frequent occurrence in urban areas. Therefore, epiphytic moss can be used for monitoring atmospheric pollutants in an immense territory and for research of atmospheric pollution in industrial centers, inhabited locations as well as assessment of atmospheric contamination in local pollution source.

Keywords: moss bioindicator, accumulation properties, NAA, AES, statistic methods

1. Introduction

Mosses concentrate a great number of heavy metals and the other trace elements from the air and precipitation in a very effective way (Coskun et al. 2011), (Grgie and Nadezdin 1990), (Onianwa 2001). They have no contact with the soil and consequently, the heterogeneous chemical composition does not practically affect them.

The morphological and physiological properties of mosses along with their widespread occurrence make them very useful to estimate surface air pollution by heavy metals (TM) and other trace elements (Burton and Agneta 1990), (Rühling 1994), (Steinnes 1989), (Steinnes et al. 1992). In conducting this type of research the main question is choosing the most suitable species of mosses based on their accumulation properties. For example, the ground species of moss (terrestrial, paludal) is used to study the transboundary transfer of atmospheric deposition of TM because they can be found in woodlands (Nikodemus et al. 2004), (Gerdol et al. 2000), (Chakrabortty and Paratkar 2006). Atmospheric pollution in urban areas can be analysed using epiphytic mosses. Epiphyte has long life cycle and grows on bark of old trees such as aspens, poplars and birches. It should be noted that the study (Chen et al. 2010), (Migaszewski et al. 2009), (Shotbolt et al. 2007) was focused on research of accumulation capacity of mosses in one systematic group – terrestrial mosses.

The purpose of this research is analysis and comparison of widespread mosses of midland: 4 species of terrestrial moss (*Pleurozium schreberi* (*Brid.*) *Mitt.*, *Dicranum polysetum Sw.*, *Ptilium cristacastrensis* (*Hedw.*) *De Not.*, *Hylocomium splendens* (Hedw.) B.S.G.), 4 species of paludal moss (Aulacomnium palustre (Brid.) Mitt., Sphagnum angustifolium (Russ. Es Russ.) C. Jens., Sphagnum squarrosum, Sphagnum centrale C. Jens. Ex H. Arnell et C. Jens.) and 2 species of epiphytic moss (Pylaisia polyantha (Hedw.) B.S.G., Orthotrichum obtusifolium Brid.).

2. Materials and methods



Fig. 1. The map of Russia, (black circle) territory of sampling mosses

2.1. Study area

The study was carried out in the town of Seversk, which is situated 56°26' N, 84° 51' E, 75-90 m above the sea level. The town is located 15 km northwest of Tomsk (Russia) (fig. 1) on the right bank terrace above flood-plain of the Tom River (fig. 2), southeast of West Siberian Plain. Population of the town is more than 100 000 inhabitants. The prevailing wind direction is south and southwest. The climate is continental with warm summers and cold winters. Snow is frequent (snowing almost 50% of winter days. The difference between the record high and the record low temperatures is 95°. The rainfall distribution is intermittent (average annual precipitation is 570 mm).



Fig. 2. The location of the sampling areas (1- Siberian Integrated Chemical Plant, 2- Tomsk Petrochemical Plant)



Fig. 3. Sampling areas and sites (1-terrestrial mosses, 2- paludal mosses, 3- epiphytic mosses)

Seversk is the town of the Siberian Group of Chemical Enterprises founded in 1954. It comprises several nuclear reactors and chemical plants for separation/ enrichment/ reprocessing of uranium and plutonium. Following an agreement signed in March 2003 between Russia and the United States to shut down Russia's three remaining reactors for plutonium production, two reactors for plutonium production located in Seversk have now been shut down.

2.2. Collection and preparation of samples

Samples of the terrestrial mosses Pleurozium schreberi, Dicranum polysetum, Ptilium cristacastrensis, Hylocomium splendens have been collected in green moss-dwarf-shrub pine forest in an area of 1 m^2 at least 80 m from roads (fig. 3 site 1). Samples of four paludal species Aulacomnium palustre, Sphagnum angustifolium, Sphagnum squarrosum, Sphagnum centrale have been collected on dwarf-shrub-sedge-sphagnum bog in a small hollow with an area 5 m^2 at least 60 m from roads (fig.3 site 2). Samples of two epiphytic species Pylaisia polyantha, Orthotrichum obtusifolium have been taken from the bark of old poplars at the height of 1,5-2 m above soil within an area 8-10 m^2 at least 50 m from roads. The sampling sites were located in the same habitat. Therefore, it can be assumed that they were exposed to the same local anthropogenic load. The moss samples were collected in dry weather conditions in August/September 2010. Sampling and sample handing was performed by using polythene gloves and paper bags. In the laboratory, the samples were carefully cleaned from dead material and litter attached. The green younger parts of the moss have been selected to make the analysis. The moss was washed with distilled water, dried to a fixed weight in a special chamber at the temperature of 85° C during 24 hours before the analysis.

To estimate accumulative properties of mosses in relation to different chemical elements, a singlefactor analysis of variance has been performed for each group of terrestrial and paludal mosses (Steel and Torrie 1980). The method of series distribution comparison over dispersions and average values for epiphytic mosses was carried out.

2.3. Measure of samples

The content of elements in the moss samples was determined by neutron activation analysis and atom emission spectrometry. Each sample was separated into 10 parallel sub-samples for statistical analysis. 32 elements in each sample have been determined by NAA and AES.

Determination of eight elements by atom emission spectrometry (AES) with inductively coupled plasma iCAP6300 DUO was carried out at Research and Analytic Centre, Tomsk Polytechnic University, Tomsk, Russia. For AES, about 0.1 g of dry mass was digested in 1 ml of H_2SO_4 during 30 min. The mixture of 4 ml of HNO₃ was heated at 200°C during 40 min and cooled at room temperature. The resulting solution was diluted with deionized water to 50 ml and reference material was added.

The neutron activation analysis was carried out at the Tomsk Research Institute, Tomsk, Russia. For NAA, the moss samples were crumbled and the probes having the mass of 0.3 g and 1 cm in diameter were pressed and packed either into aluminum foil for their long-term irradiation. Samples have been irradiated in IRT-T reactor channels (thermal neutron flux density of 5.5 x 10^{13} n cm⁻² s⁻¹). To determine elements yielding long-lived isotopes, reference materials have been packed together with 20-25 samples in a container and were irradiated for 5 hours. Then reference materials and samples were re-packed and measurements have been made thrice after 7, 30 and 60 days of decay. The duration of measurements was 10, 20 and 30 min, respectively. The measurement of the chemical element specific activities in the probes was made by the semiconductor gamma spectrometer. The gamma spectra of induced activity were analyzed by means of "Genia-2000" software program developed by the "CONBERRA".

To ensure quality control, contents of elements yielding long-living isotopes have been determined using certified reference materials: INCT-OBTL-5

(Institute of Nuclear Chemistry and Technology Oriental Basma Tobacco Leaves) and IAEA/V-10 Hay (International Atomic Energy Agency).

3. Results and discussion

The concentrations of chemical elements for each moss species are normally distributed. The single-factor analysis is tested by comparing the F test statistic:

 $F = \frac{\sigma_F^2}{\sigma_0^2},$ Where: $\sigma_F^2 = \frac{\sum_{i=1}^r (\overline{x}_i - \overline{x})^2}{r-1}$ and $\sigma_0^2 = \frac{\sum_{i=1}^r \sum_{j=1}^n (x_j - \overline{x}_i)^2}{n-r}$ - are intergroup and intragroup

dispersions accordingly figured out upon one degree of freedom.

Where: \overline{x} -overall mean, \overline{x}_i - group average, n_i - number of tests in group, n- total number of tests, r-a number of groups. This F-statistic follows the Fdistribution with r-1, n-*r* degrees of freedom under the null hypothesis. The calculated F-ratios significantly exceed the critical values ($F_{cr} = 2.8$ for elements determined by NAA; $F_{cr} = 3.09$ for elements determined by AES, the probability is equal to 0.95 (Steel and Torrie 1980) for all chemical elements. This goes to prove different accumulation capacity of both terrestrial and paludal moss species. Hence, in the course of the only investigation, the use of various terrestrial mosses collected from one area is incorrect.

The comparison of average values of concentrations for epiphitic mosses with a t-test has been carried out for the elements whose ratio of variances does not exceed the critical value. The analysis showed that only 3 elements – Th, Cs, Co – do not reveal any significant differences in the compared variances and average values. Thus, compared populations are heterogeneous, namely the differences of accumulation properties for various epiphytic moss species in most chemical elements.

To determine the extent of accumulation capacity, the average value concentrations of all chemical elements in an increasing order have been ranged (Manoukian 1986). Each concentration of a chemical element is assigned a rank. The number of ranks is equal to the amount of the moss species studied. In this instance, there are 10 ranks. The concentrations of Mo, Cr, Zn and corresponding concentration ranks are presented as an example in Table 1. The rank sums of all the chemical elements for each moss species have been calculated (Table 2).

Table 1. Concentration values for Mo, Cr, Zn in various mosses species (mg/kg) and their corresponding ranks

Encoing of magaag	Мо		Cr		Zn	
Species of mosses	value	rank	value	rank	value	rank
Dicranum polysetum	1.6	6	16	8	55	4
Aulacomnium palustre	1.9	8	9.8	4	67	7
Ptilium crista-castrensis	1.1	3	12	7	56	5
Sphagnum centrale	0.45	2	11	5	46	2
Hylocomium splendens	1.1	4	8.4	3	38	1
Sphagnum angustifolium	1.7	7	8.1	1	56	6
Sphagnum squarrosum	1.1	5	11	6	126	8
Pleurozium schreberi	0.45	1	8.2	2	51	3
Pylasia polyantha	72	10	92	10	307	9
Orthotrichum obtusifolium	4.1	9	50	9	542	10

The obtained data is well explained by the anatomical and morphological moss structure. For instance, the highest rank sum among terrestrial mosses was obtained for *Dicranum polysetum*, which has ample rhizoid felt and dense structure of bunches that probably forwards the retention of polluting components. Therefore, this species of moss has the

highest accumulating capacity in comparison to other studied terrestrial mosses. Moss species *Hylocomium splendens* has the smallest rank sum, i.e. this moss species has the lowest accumulation capacity due to loose structure of its bunches and location of branches on the bine.

Table 2. Rank sums for the various terrestrial mosses

Mosses species	Rank sum	Mosses species	Rank sum
Aulacomnium palustre	108	Ptilium crista-castrensis	104
Sphagnum centrale	57	Hylocomium splendens	62
Sphagnum angustifolium	59	Pleurozium schreberi	75
Sphagnum squarrosum	95	Pylaisia polyantha	173
Dicranum polysetum	142	Orthotrichum obtusifolium	170

Moss species *Hylocomium splendens* and *Pleurozium schreberi* used for monitoring in Europe (Berg and Steinnes 1997), (Culicov et al. 2004) have

approximately similar rank sum. It should be noted that the mosses concerned have different accumulation capacities in relation to the various elements. When concentration values of elements are compared, it becomes evident that *Pleurozium* schreberi accumulates As, Ba, Ni, Hf, Zn, Na 1.5-2 times more than *Hylocomium* splendens and *Hylocomium* splendens accumulates Sm, Mo, Th, Hg - 1.5-2 times more than *Pleurozium* schreberi as well (table 3).

Aulacomnium palustre with ample rhizoid felt along the whole bine has the highest accumulation capacity among paludal mosses. Sphagnum centrale and Sphagnum angustifolium have approximately the same rank sum, because they relate to the same genus and possess similar anatomical-morphological and physiological characteristics (Tyler 1990).

Table 3 contains the average values of element concentrations obtained for two epiphytic

polyantha, mosses (Pylaisia Orthotrichum obtusifolium) and the terrestrial moss (Dicranum polysetum) having the highest accumulating abilities. Table 3 shows that epiphytic mosses accumulate and retain elements to a greater extent in comparison to terrestrial mosses. Thus, the advantage of using epiphytic mosses as biomonitors is not only the absence of the influence of soil on the content of elements, which ensures high comparability of the obtained results in different areas, but also in their high accumulation capacity compared to terrestrial mosses. Besides, the use of epiphytic mosses enables analysis of atmospheric pollution in industrial centers and inhabited areas. This is practically impossible when terrestrial and paludal mosses are used.

 Table 3.
 Values of concentrations and standard errors in mosses (1- Dicranium polysetum, 2- Ptilium crista-castrensis, 3-Hylocomium splendeus, 4- Pleurozium schreberi, 5- Sphagnum central, 6- Sphagnum angustifalium, 7-Sphagnum squarrosum, 8- Aulacomnium palustre, 9- Pylaisia polyantha, 10- Orthotrichum obtusifolium, mg/kg

1 1	Al*	As	Ba	Br	Ca	Cd*	Со	Cr
1	1620±167	0.54±0.04	89±4	3.2±0.2	10535±450	0.45±0.25	1.3±0.04	16±0.7
	1350±344	0.31 ± 0.034	69±3	3.3±0.3	10012 ± 362	0.33±0.81	0.98±0.04	12 ± 0.5
	1340±55	0.05±0.004	67±5	2.7±0.2	7397±337	0.31±0.28	0.79±0.06	8.4±0.3
	1148.9±111	0.11±0.007	89±11	2.8±0.2	8068±559	0.44±0.28	0.83±0.04	8.2±0.4
	994±53	0.04±0.006	57±4	3.8±0.2	5727±289	0.13±0.02	0.71±0.04	11±1.0
	805±20	0.10±0.01	107±7	3.2±0.2	5560±253	0.07±0.01	0.71±0.03	8.1±0.4
	921±54	0.13±0.01	132±8	3.2±0.4	9515±983	0.17±0.02	0.87±0.03	11±1.4
8	1398±208	0.61±0.03	54±3	3.1±0.1	7629±385	0.38±0.09	1.2±0.04	9.8±0.2
9	-	0.92±0.12	290±50	5.2±0.9	12444±343	-	2.2±0.1	92±5
10	-	0.73±0.03	342±19	18±2.0	13660±1445	-	1.9±0.2	50±3
	Cs	Cu*	Eu	Fe	Hf	Hg*	La	Mg*
1	0.51±0.02	11±0.4	0.069 ± 0.002	2800±80	0.55±0.02	0.015±0.0007	4.7±0.2	1531±87
2	0.29±0.01	9.7±1.9	0.05 ± 0.002	2099±116	0.21±0.01	0.03 ± 0.001	2.2±0.1	1560±119
-	0.27±0.02	14±3.2	0.061±0.003		1.05 ± 0.06	0.027±0.001	2.6±0.1	1327±63
4	0.38±0.02	11±1.1	0.066 ± 0.004	2078±138	0.38±0.03	0.017±0.002	2.3±0.1	1253±55
5	0.19±0.01	7.5±0.3	0.058 ± 0.002	1526±96	0.38±0.02	0.025±0.001	2.7±0.1	1144±105
6	0.24±0.01	5.7±0.5	0.045±0.002	1666±75	0.38±0.03	0.014±0.002	2.3±0.1	1114±75
7	0.30±0.02	6.3±0.4	0.066 ± 0.004	1955±88	0.43±0.06	0.011±0.002	2.6±0.1	2049±75
8	0.34±0.01	9.9±0.5	0.073±0.005	2669±96	0.46±0.03	0.028 ± 0.002	4.4±0.2	1551±104
9	$0.44{\pm}0.09$	-	-	15703±257	-	-	8.1±0.3	-
10	0.56 ± 0.08	-	_	5916±228			7.1±0.3	_
10	5.20-0.00	-	-	3710 ± 220	-	-	7.1±0.5	-
	Mn*	Mo	- Na	Ni*	- Pb*	- Rb	Sb	- Sc
1	Mn* 520±21	Mo 1.6±0.2	1174±36	Ni* 13±0.7	Pb* 6.9±1.6	- Rb 18±0.9		0.86±0.03
1	Mn*	1.6±0.2 1.0±0.1	1174±36 832±53	Ni* 13±0.7 10±0.6	6.9±1.6 6.4±1.3	18±0.9 20±1	Sb 0.71±0.05 0.48±0.03	0.86±0.03 0.62±0.03
1 2	Mn* 520±21	1.6±0.2 1.0±0.1 1.0±0.1	1174±36 832±53 229±14	Ni* 13±0.7	6.9±1.6 6.4±1.3 6.6±2.2	18±0.9	Sb 0.71±0.05 0.48±0.03 0.25±0.01	0.86±0.03 0.62±0.03 0.55±0.03
1 2 3	Mn* 520±21 509±52	1.6±0.2 1.0±0.1 1.0±0.1 0.45±0.05	1174±36 832±53 229±14 785±55	Ni* 13±0.7 10±0.6 4.8±0.3 10±0.9	6.9±1.6 6.4±1.3 6.6±2.2 6.4±1.4	18±0.9 20±1 13±0.8 17±1.3	Sb 0.71±0.05 0.48±0.03 0.25±0.01 0.32±0.02	0.86±0.03 0.62±0.03 0.55±0.03 0.54±0.02
1 2 3 4 5	Mn* 520±21 509±52 415±10 448±2 290±6	1.6±0.2 1.0±0.1 1.0±0.1	$ \begin{array}{r} 1174 \pm 36 \\ 832 \pm 53 \\ 229 \pm 14 \\ 785 \pm 55 \\ 1104 \pm 75 \\ \end{array} $	Ni* 13±0.7 10±0.6 4.8±0.3 10±0.9 4.8±0.2	6.9±1.6 6.4±1.3 6.6±2.2 6.4±1.4 3.5±2.0	18±0.9 20±1 13±0.8 17±1.3 6.8±0.4	Sb 0.71±0.05 0.48±0.03 0.25±0.01 0.32±0.02 0.13±0.01	0.86±0.03 0.62±0.03 0.55±0.03 0.54±0.02 0.45±0.02
1 2 3 4 5 6	Mn* 520±21 509±52 415±10 448±2 290±6 390±33	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.05\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1 \end{array}$	1174±36 832±53 229±14 785±55 1104±75 474±29	Ni* 13±0.7 10±0.6 4.8±0.3 10±0.9 4.8±0.2 3.9±0.4	6.9±1.6 6.4±1.3 6.6±2.2 6.4±1.4 3.5±2.0 1.4±0.5	18±0.9 20±1 13±0.8 17±1.3 6.8±0.4 10±1.1	Sb 0.71±0.05 0.48±0.03 0.25±0.01 0.32±0.02 0.13±0.01 0.31±0.02	$\begin{array}{c} 0.86{\pm}0.03\\ \hline 0.62{\pm}0.03\\ \hline 0.55{\pm}0.03\\ \hline 0.54{\pm}0.02\\ \hline 0.45{\pm}0.02\\ \hline 0.49{\pm}0.02\\ \end{array}$
1 2 3 4 5 6	Mn* 520±21 509±52 415±10 448±2 290±6	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.05\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ \end{array}$	$ \begin{array}{r} 1174 \pm 36 \\ 832 \pm 53 \\ 229 \pm 14 \\ 785 \pm 55 \\ 1104 \pm 75 \\ \end{array} $	Ni* 13±0.7 10±0.6 4.8±0.3 10±0.9 4.8±0.2 3.9±0.4 9.8±0.8	6.9±1.6 6.4±1.3 6.6±2.2 6.4±1.4 3.5±2.0	$ \begin{array}{r} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ \end{array} $	Sb 0.71±0.05 0.48±0.03 0.25±0.01 0.32±0.02 0.13±0.01 0.31±0.02 0.16±0.01	$\begin{array}{c} 0.86 {\pm} 0.03 \\ 0.62 {\pm} 0.03 \\ 0.55 {\pm} 0.03 \\ 0.54 {\pm} 0.02 \\ 0.45 {\pm} 0.02 \\ 0.49 {\pm} 0.02 \\ 0.58 {\pm} 0.03 \end{array}$
1 2 3 4 5 6 7	Mn* 520±21 509±52 415±10 448±2 290±6 390±33	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.05\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ 1.9{\pm}0.1\\ \end{array}$	$ \begin{array}{r} 1174 \pm 36 \\ 832 \pm 53 \\ 229 \pm 14 \\ 785 \pm 55 \\ 1104 \pm 75 \\ 474 \pm 29 \\ 706 \pm 76 \\ 996 \pm 46 \\ \end{array} $	Ni* 13±0.7 10±0.6 4.8±0.3 10±0.9 4.8±0.2 3.9±0.4 9.8±0.8 7.8±0.4	6.9±1.6 6.4±1.3 6.6±2.2 6.4±1.4 3.5±2.0 1.4±0.5	$\begin{array}{c} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ 10\pm0.8\\ \end{array}$	$\begin{array}{c} \textbf{Sb} \\ 0.71 \pm 0.05 \\ 0.48 \pm 0.03 \\ 0.25 \pm 0.01 \\ 0.32 \pm 0.02 \\ 0.13 \pm 0.01 \\ 0.31 \pm 0.02 \\ 0.16 \pm 0.01 \\ 0.45 \pm 0.03 \end{array}$	$\begin{array}{c} 0.86{\pm}0.03\\ 0.62{\pm}0.03\\ 0.55{\pm}0.03\\ 0.54{\pm}0.02\\ 0.45{\pm}0.02\\ 0.49{\pm}0.02\\ 0.58{\pm}0.03\\ 0.74{\pm}0.03\\ \end{array}$
1 2 3 4 5 6 7 8 9	Mn* 520±21 509±52 415±10 448±2 290±6 390±33 504±20	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.05\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ 1.9{\pm}0.1\\ 72{\pm}7.0\\ \end{array}$	$ \begin{array}{r} 1174 \pm 36 \\ 832 \pm 53 \\ 229 \pm 14 \\ 785 \pm 55 \\ 1104 \pm 75 \\ 474 \pm 29 \\ 706 \pm 76 \\ 996 \pm 46 \\ 3075 \pm 254 \\ \end{array} $	Ni* 13±0.7 10±0.6 4.8±0.3 10±0.9 4.8±0.2 3.9±0.4 9.8±0.8 7.8±0.4 59±3	6.9±1.6 6.4±1.3 6.6±2.2 6.4±1.4 3.5±2.0 1.4±0.5 3.1±0.7	$\begin{array}{c} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ 10\pm0.8\\ 24\pm3\\ \end{array}$	$\begin{array}{c} \textbf{Sb} \\ 0.71 \pm 0.05 \\ 0.48 \pm 0.03 \\ 0.25 \pm 0.01 \\ 0.32 \pm 0.02 \\ 0.13 \pm 0.01 \\ 0.31 \pm 0.02 \\ 0.16 \pm 0.01 \\ 0.45 \pm 0.03 \\ 1.2 \pm 0.05 \end{array}$	$\begin{array}{c} 0.86{\pm}0.03\\ \hline 0.62{\pm}0.03\\ \hline 0.55{\pm}0.03\\ \hline 0.54{\pm}0.02\\ \hline 0.45{\pm}0.02\\ \hline 0.49{\pm}0.02\\ \hline 0.58{\pm}0.03\\ \hline 0.74{\pm}0.03\\ \hline 1.6{\pm}0.08\\ \end{array}$
1 2 3 4 5 6 7	Mn* 520±21 509±52 415±10 448±2 290±6 390±33 504±20 273±20 - -	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.05\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ 1.9{\pm}0.1\\ 72{\pm}7.0\\ 4.1{\pm}0.5 \end{array}$	$\begin{array}{c} 1174{\pm}36\\ 832{\pm}53\\ 229{\pm}14\\ 785{\pm}55\\ 1104{\pm}75\\ 474{\pm}29\\ 706{\pm}76\\ 996{\pm}46\\ 3075{\pm}254\\ 2476{\pm}156\\ \end{array}$	Ni* 13±0.7 10±0.6 4.8±0.3 10±0.9 4.8±0.2 3.9±0.4 9.8±0.8 7.8±0.4 59±3 85±5	6.9±1.6 6.4±1.3 6.6±2.2 6.4±1.4 3.5±2.0 1.4±0.5 3.1±0.7 5.3±1.9 -	$\begin{array}{c} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ 10\pm0.8\\ 24\pm3\\ 18\pm2 \end{array}$	$\begin{array}{c} \textbf{Sb} \\ 0.71 \pm 0.05 \\ 0.48 \pm 0.03 \\ 0.25 \pm 0.01 \\ 0.32 \pm 0.02 \\ 0.13 \pm 0.01 \\ 0.31 \pm 0.02 \\ 0.16 \pm 0.01 \\ 0.45 \pm 0.03 \\ 1.2 \pm 0.05 \\ 0.91 \pm 0.08 \end{array}$	$\begin{array}{c} 0.86{\pm}0.03\\ \hline 0.62{\pm}0.03\\ \hline 0.55{\pm}0.03\\ \hline 0.54{\pm}0.02\\ \hline 0.45{\pm}0.02\\ \hline 0.49{\pm}0.02\\ \hline 0.58{\pm}0.03\\ \hline 0.74{\pm}0.03\\ \hline 1.6{\pm}0.08\\ \hline 1.8{\pm}0.03\\ \end{array}$
1 2 3 4 5 6 7 8 9 10	Mn* 520±21 509±52 415±10 448±2 290±6 390±33 504±20 273±20 - - Se*	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.05\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ 1.9{\pm}0.1\\ 72{\pm}7.0\\ 4.1{\pm}0.5\\ \mathbf{Si}* \end{array}$	$\begin{array}{c} 1174 \pm 36 \\ 832 \pm 53 \\ 229 \pm 14 \\ 785 \pm 55 \\ 1104 \pm 75 \\ 474 \pm 29 \\ 706 \pm 76 \\ 996 \pm 46 \\ 3075 \pm 254 \\ 2476 \pm 156 \\ \mathbf{Sm} \end{array}$	$\begin{array}{r} Ni^{*} \\ 13\pm0.7 \\ 10\pm0.6 \\ 4.8\pm0.3 \\ 10\pm0.9 \\ 4.8\pm0.2 \\ 3.9\pm0.4 \\ 9.8\pm0.8 \\ 7.8\pm0.4 \\ 59\pm3 \\ 85\pm5 \\ Sr \end{array}$	6.9±1.6 6.4±1.3 6.6±2.2 6.4±1.4 3.5±2.0 1.4±0.5 3.1±0.7 5.3±1.9 - - Th	$\begin{array}{c} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ 10\pm0.8\\ 24\pm3\\ 18\pm2\\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ $	Sb 0.71±0.05 0.48±0.03 0.25±0.01 0.32±0.02 0.13±0.01 0.31±0.02 0.16±0.01 0.45±0.03 1.2±0.05 0.91±0.08 V*	$\begin{array}{c} 0.86{\pm}0.03\\ \hline 0.62{\pm}0.03\\ \hline 0.55{\pm}0.03\\ \hline 0.54{\pm}0.02\\ \hline 0.45{\pm}0.02\\ \hline 0.49{\pm}0.02\\ \hline 0.58{\pm}0.03\\ \hline 0.74{\pm}0.03\\ \hline 1.6{\pm}0.08\\ \hline 1.8{\pm}0.03\\ \hline \mathbf{Zn} \end{array}$
1 2 3 4 5 6 7 8 9 10 1	Mn* 520±21 509±52 415±10 448±2 290±6 390±33 504±20 273±20 - - Se* 4.5±1.3	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.05\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ 1.9{\pm}0.1\\ 72{\pm}7.0\\ 4.1{\pm}0.5\\ \hline {\bf Si}{*}\\ 1271{\pm}85 \end{array}$	$\begin{array}{c} 1174 \pm 36 \\ 832 \pm 53 \\ 229 \pm 14 \\ 785 \pm 55 \\ 1104 \pm 75 \\ 474 \pm 29 \\ 706 \pm 76 \\ 996 \pm 46 \\ 3075 \pm 254 \\ 2476 \pm 156 \\ \hline \mathbf{Sm} \\ 1.3 \pm 0.05 \\ \end{array}$	$\begin{array}{r} Ni^{*} \\ 13\pm0.7 \\ 10\pm0.6 \\ 4.8\pm0.3 \\ 10\pm0.9 \\ 4.8\pm0.2 \\ 3.9\pm0.4 \\ 9.8\pm0.8 \\ 7.8\pm0.4 \\ 59\pm3 \\ 85\pm5 \\ \hline Sr \\ 68\pm5 \end{array}$	6.9±1.6 6.4±1.3 6.6±2.2 6.4±1.4 3.5±2.0 1.4±0.5 3.1±0.7 5.3±1.9 - Th 0.98±0.03	$\begin{array}{c} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ 10\pm0.8\\ 24\pm3\\ 18\pm2\\ \hline {\bf Ti}*\\ 129\pm14\\ \end{array}$	$\begin{array}{c} \textbf{Sb} \\ 0.71 \pm 0.05 \\ 0.48 \pm 0.03 \\ 0.25 \pm 0.01 \\ 0.32 \pm 0.02 \\ 0.13 \pm 0.01 \\ 0.31 \pm 0.02 \\ 0.16 \pm 0.01 \\ 0.45 \pm 0.03 \\ 1.2 \pm 0.05 \\ 0.91 \pm 0.08 \\ \textbf{V*} \\ 3.6 \pm 0.6 \end{array}$	$\begin{array}{c} 0.86{\pm}0.03\\ 0.62{\pm}0.03\\ 0.55{\pm}0.03\\ 0.55{\pm}0.02\\ 0.45{\pm}0.02\\ 0.45{\pm}0.02\\ 0.49{\pm}0.02\\ 0.58{\pm}0.03\\ 0.74{\pm}0.03\\ 1.6{\pm}0.08\\ 1.8{\pm}0.03\\ \mathbf{Zn}\\ 55{\pm}2 \end{array}$
1 2 3 4 5 6 7 8 9 10 1 2	Mn* 520±21 509±52 415±10 448±2 290±6 390±33 504±20 273±20 - - Se* 4.5±1.3 5.3±0.7	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.05\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ 1.9{\pm}0.1\\ 72{\pm}7.0\\ 4.1{\pm}0.5\\ \textbf{Si*}\\ 1271{\pm}85\\ 1154{\pm}178 \end{array}$	$\begin{array}{c} 1174{\pm}36\\ 832{\pm}53\\ 229{\pm}14\\ 785{\pm}55\\ 1104{\pm}75\\ 474{\pm}29\\ 706{\pm}76\\ 996{\pm}46\\ 3075{\pm}254\\ 2476{\pm}156\\ {\color{black}{\textbf{Sm}}}\\ 1.3{\pm}0.05\\ 0.55{\pm}0.02\\ \end{array}$	$\begin{array}{r} Ni^{*} \\ 13\pm0.7 \\ 10\pm0.6 \\ 4.8\pm0.3 \\ 10\pm0.9 \\ 4.8\pm0.2 \\ 3.9\pm0.4 \\ 9.8\pm0.8 \\ 7.8\pm0.4 \\ 59\pm3 \\ 85\pm5 \\ \hline Sr \\ 68\pm5 \\ 59\pm3 \\ \end{array}$	$\begin{array}{c} 6.9 \pm 1.6 \\ 6.4 \pm 1.3 \\ 6.6 \pm 2.2 \\ 6.4 \pm 1.4 \\ 3.5 \pm 2.0 \\ 1.4 \pm 0.5 \\ 3.1 \pm 0.7 \\ 5.3 \pm 1.9 \\ - \\ - \\ \hline \mathbf{Th} \\ 0.98 \pm 0.03 \\ 0.59 \pm 0.02 \end{array}$	$\begin{array}{c} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ 10\pm0.8\\ 24\pm3\\ 18\pm2\\ \hline {\bf Ti}*\\ 129\pm14\\ 115\pm32\\ \end{array}$	$\begin{array}{c} \textbf{Sb} \\ 0.71 \pm 0.05 \\ 0.48 \pm 0.03 \\ 0.25 \pm 0.01 \\ 0.32 \pm 0.02 \\ 0.13 \pm 0.01 \\ 0.31 \pm 0.02 \\ 0.16 \pm 0.01 \\ 0.45 \pm 0.03 \\ 1.2 \pm 0.05 \\ 0.91 \pm 0.08 \\ \textbf{V*} \\ 3.6 \pm 0.6 \\ 2.9 \pm 0.8 \end{array}$	$\begin{array}{c} 0.86\pm 0.03 \\ \hline 0.62\pm 0.03 \\ \hline 0.55\pm 0.03 \\ \hline 0.55\pm 0.02 \\ \hline 0.45\pm 0.02 \\ \hline 0.49\pm 0.02 \\ \hline 0.49\pm 0.02 \\ \hline 0.58\pm 0.03 \\ \hline 0.74\pm 0.03 \\ \hline 1.6\pm 0.08 \\ \hline 1.8\pm 0.03 \\ \hline \mathbf{Zn} \\ \hline 55\pm 2 \\ \hline 56\pm 2 \end{array}$
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Mn* 520±21 509±52 415±10 448±2 290±6 390±33 504±20 273±20 - - Se* 4.5±1.3 5.3±0.7 4.8±0.7	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.05\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ 1.9{\pm}0.1\\ 72{\pm}7.0\\ 4.1{\pm}0.5\\ {\bf Si}{*}\\ 1271{\pm}85\\ 1154{\pm}178\\ 1166{\pm}100\\ \end{array}$	$\begin{array}{c} 1174{\pm}36\\ 832{\pm}53\\ 229{\pm}14\\ 785{\pm}55\\ 1104{\pm}75\\ 474{\pm}29\\ 706{\pm}76\\ 996{\pm}46\\ 3075{\pm}254\\ 2476{\pm}156\\ {\color{red}{\textbf{Sm}}}\\ 1.3{\pm}0.05\\ 0.55{\pm}0.02\\ 0.62{\pm}0.06\\ \end{array}$	$\begin{array}{r} Ni^{*} \\ 13\pm0.7 \\ 10\pm0.6 \\ 4.8\pm0.3 \\ 10\pm0.9 \\ 4.8\pm0.2 \\ 3.9\pm0.4 \\ 9.8\pm0.8 \\ 7.8\pm0.4 \\ 59\pm3 \\ 85\pm5 \\ \hline Sr \\ 68\pm5 \\ 59\pm3 \\ 34\pm2 \\ \end{array}$	$\begin{array}{c} 6.9 \pm 1.6 \\ 6.4 \pm 1.3 \\ 6.6 \pm 2.2 \\ 6.4 \pm 1.4 \\ 3.5 \pm 2.0 \\ 1.4 \pm 0.5 \\ 3.1 \pm 0.7 \\ 5.3 \pm 1.9 \\ - \\ \hline \\ \mathbf{Th} \\ 0.98 \pm 0.03 \\ 0.59 \pm 0.02 \\ 0.74 \pm 0.04 \end{array}$	$\begin{array}{c} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ 10\pm0.8\\ 24\pm3\\ 18\pm2\\ \hline {\bf Ti}*\\ 129\pm14\\ 115\pm32\\ 100\pm3\\ \end{array}$	$\begin{array}{c} \textbf{Sb} \\ 0.71 \pm 0.05 \\ 0.48 \pm 0.03 \\ 0.25 \pm 0.01 \\ 0.32 \pm 0.02 \\ 0.13 \pm 0.01 \\ 0.31 \pm 0.02 \\ 0.16 \pm 0.01 \\ 0.45 \pm 0.03 \\ 1.2 \pm 0.05 \\ 0.91 \pm 0.08 \\ \textbf{V*} \\ 3.6 \pm 0.6 \\ 2.9 \pm 0.8 \\ 3.1 \pm 0.3 \end{array}$	$\begin{array}{c} 0.86{\pm}0.03\\ 0.62{\pm}0.03\\ 0.55{\pm}0.03\\ 0.55{\pm}0.02\\ 0.45{\pm}0.02\\ 0.45{\pm}0.02\\ 0.49{\pm}0.02\\ 0.58{\pm}0.03\\ 0.74{\pm}0.03\\ 1.6{\pm}0.08\\ 1.8{\pm}0.03\\ \hline \mathbf{Zn}\\ 55{\pm}2\\ 56{\pm}2\\ 38{\pm}2\\ \end{array}$
$ \begin{array}{c} 1\\ 2\\ 3\\ 4\\ 5\\ 6\\ 7\\ 8\\ 9\\ 10\\ 1\\ 2\\ 3\\ 4\\ 4\\ \end{array} $	Mn* 520±21 509±52 415±10 448±2 290±6 390±33 504±20 273±20 - - Se* 4.5±1.3 5.3±0.7 4.8±0.7 5.2±1.1	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.05\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ 1.9{\pm}0.1\\ 72{\pm}7.0\\ 4.1{\pm}0.5\\ {\bf Si}{\bf *}\\ 1271{\pm}85\\ 1154{\pm}178\\ 1166{\pm}100\\ 1046{\pm}144\\ \end{array}$	$\begin{array}{c} 1174{\pm}36\\ 832{\pm}53\\ 229{\pm}14\\ 785{\pm}55\\ 1104{\pm}75\\ 474{\pm}29\\ 706{\pm}76\\ 996{\pm}46\\ 3075{\pm}254\\ 2476{\pm}156\\ {\color{black}{\mathbf{Sm}}}\\ 1.3{\pm}0.05\\ 0.55{\pm}0.02\\ 0.62{\pm}0.06\\ 0.39{\pm}0.03\\ \end{array}$	$\begin{array}{r} Ni^{*} \\ 13\pm0.7 \\ 10\pm0.6 \\ 4.8\pm0.3 \\ 10\pm0.9 \\ 4.8\pm0.2 \\ 3.9\pm0.4 \\ 9.8\pm0.8 \\ 7.8\pm0.4 \\ 59\pm3 \\ 85\pm5 \\ \hline Sr \\ 68\pm5 \\ 59\pm3 \\ 34\pm2 \\ 46\pm4 \\ \end{array}$	$\begin{array}{c} 6.9 \pm 1.6 \\ 6.4 \pm 1.3 \\ 6.6 \pm 2.2 \\ 6.4 \pm 1.4 \\ 3.5 \pm 2.0 \\ 1.4 \pm 0.5 \\ 3.1 \pm 0.7 \\ 5.3 \pm 1.9 \\ - \\ \hline \mathbf{Th} \\ 0.98 \pm 0.03 \\ 0.59 \pm 0.02 \\ 0.74 \pm 0.04 \\ 0.54 \pm 0.03 \end{array}$	$\begin{array}{c} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ 10\pm0.8\\ 24\pm3\\ 18\pm2\\ \hline {\bf Ti}*\\ 129\pm14\\ 115\pm32\\ 100\pm3\\ 86\pm5 \end{array}$	$\begin{array}{c} \textbf{Sb} \\ 0.71 \pm 0.05 \\ 0.48 \pm 0.03 \\ 0.25 \pm 0.01 \\ 0.32 \pm 0.02 \\ 0.13 \pm 0.01 \\ 0.31 \pm 0.02 \\ 0.16 \pm 0.01 \\ 0.45 \pm 0.03 \\ 1.2 \pm 0.05 \\ 0.91 \pm 0.08 \\ \textbf{V*} \\ 3.6 \pm 0.6 \\ 2.9 \pm 0.8 \\ 3.1 \pm 0.3 \\ 2.7 \pm 0.2 \end{array}$	$\begin{array}{c} 0.86{\pm}0.03\\ 0.62{\pm}0.03\\ 0.55{\pm}0.03\\ 0.55{\pm}0.02\\ 0.45{\pm}0.02\\ 0.49{\pm}0.02\\ 0.58{\pm}0.03\\ 0.74{\pm}0.03\\ 1.6{\pm}0.08\\ 1.8{\pm}0.03\\ \hline \textbf{Zn}\\ 55{\pm}2\\ 56{\pm}2\\ 38{\pm}2\\ 51{\pm}3\\ \end{array}$
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Mn* 520±21 509±52 415±10 448±2 290±6 390±33 504±20 273±20 - - Se* 4.5±1.3 5.3±0.7 4.8±0.7 5.2±1.1 4.3±1.0	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.05\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ 1.9{\pm}0.1\\ 72{\pm}7.0\\ 4.1{\pm}0.5\\ {\bf Si}{\bf *}\\ 1271{\pm}85\\ 1154{\pm}178\\ 1166{\pm}100\\ 1046{\pm}144\\ 803{\pm}138\\ \end{array}$	$\begin{array}{c} 1174{\pm}36\\ 832{\pm}53\\ 229{\pm}14\\ 785{\pm}55\\ 1104{\pm}75\\ 474{\pm}29\\ 706{\pm}76\\ 996{\pm}46\\ 3075{\pm}254\\ 2476{\pm}156\\ {\color{black}{\mathbf{Sm}}}\\ 1.3{\pm}0.05\\ 0.55{\pm}0.02\\ 0.62{\pm}0.06\\ 0.39{\pm}0.03\\ 0.64{\pm}0.045\\ \end{array}$	$\begin{array}{r} Ni^{*} \\ 13\pm0.7 \\ 10\pm0.6 \\ 4.8\pm0.3 \\ 10\pm0.9 \\ 4.8\pm0.2 \\ 3.9\pm0.4 \\ 9.8\pm0.8 \\ 7.8\pm0.4 \\ 59\pm3 \\ 85\pm5 \\ \hline Sr \\ 68\pm5 \\ 59\pm3 \\ 34\pm2 \\ 46\pm4 \\ 24\pm2 \\ \end{array}$	$\begin{array}{c} 6.9 \pm 1.6 \\ 6.4 \pm 1.3 \\ 6.6 \pm 2.2 \\ 6.4 \pm 1.4 \\ 3.5 \pm 2.0 \\ 1.4 \pm 0.5 \\ 3.1 \pm 0.7 \\ 5.3 \pm 1.9 \\ - \\ \hline \mathbf{Th} \\ 0.98 \pm 0.03 \\ 0.59 \pm 0.02 \\ 0.74 \pm 0.04 \\ 0.54 \pm 0.03 \\ 0.68 \pm 0.04 \end{array}$	$\begin{array}{c} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ 10\pm0.8\\ 24\pm3\\ 18\pm2\\ \hline \mathbf{Ti*}\\ 129\pm14\\ 115\pm32\\ 100\pm3\\ 86\pm5\\ 72\pm4\\ \end{array}$	$\begin{array}{c} \textbf{Sb} \\ 0.71 \pm 0.05 \\ 0.48 \pm 0.03 \\ 0.25 \pm 0.01 \\ 0.32 \pm 0.02 \\ 0.13 \pm 0.01 \\ 0.31 \pm 0.02 \\ 0.16 \pm 0.01 \\ 0.45 \pm 0.03 \\ 1.2 \pm 0.05 \\ 0.91 \pm 0.08 \\ \textbf{V*} \\ 3.6 \pm 0.6 \\ 2.9 \pm 0.8 \\ 3.1 \pm 0.3 \\ 2.7 \pm 0.2 \\ 2.0 \pm 0.3 \end{array}$	$\begin{array}{c} 0.86{\pm}0.03\\ 0.62{\pm}0.03\\ 0.55{\pm}0.03\\ 0.55{\pm}0.02\\ 0.45{\pm}0.02\\ 0.49{\pm}0.02\\ 0.58{\pm}0.03\\ 0.74{\pm}0.03\\ 1.6{\pm}0.08\\ 1.8{\pm}0.03\\ \hline \textbf{Zn}\\ 55{\pm}2\\ 56{\pm}2\\ 38{\pm}2\\ 51{\pm}3\\ 46{\pm}3\\ \end{array}$
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Mn* 520±21 509±52 415±10 448±2 290±6 390±33 504±20 273±20 - - Se* 4.5±1.3 5.3±0.7 4.8±0.7 5.2±1.1 4.3±1.0 4.0±1.0	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.05\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ 1.9{\pm}0.1\\ 72{\pm}7.0\\ 4.1{\pm}0.5\\ {\bf Si^*}\\ 1271{\pm}85\\ 1154{\pm}178\\ 1166{\pm}100\\ 1046{\pm}144\\ 803{\pm}138\\ 643{\pm}51\\ \end{array}$	$\begin{array}{c} 1174{\pm}36\\ 832{\pm}53\\ 229{\pm}14\\ 785{\pm}55\\ 1104{\pm}75\\ 474{\pm}29\\ 706{\pm}76\\ 996{\pm}46\\ 3075{\pm}254\\ 2476{\pm}156\\ \textbf{Sm}\\ 1.3{\pm}0.05\\ 0.55{\pm}0.02\\ 0.62{\pm}0.06\\ 0.39{\pm}0.03\\ 0.64{\pm}0.045\\ 0.51{\pm}0.029\\ \end{array}$	$\begin{array}{r} Ni^{*} \\ 13\pm0.7 \\ 10\pm0.6 \\ 4.8\pm0.3 \\ 10\pm0.9 \\ 4.8\pm0.2 \\ 3.9\pm0.4 \\ 9.8\pm0.8 \\ 7.8\pm0.4 \\ 59\pm3 \\ 85\pm5 \\ \hline Sr \\ 68\pm5 \\ 59\pm3 \\ 34\pm2 \\ 46\pm4 \\ 24\pm2 \\ 29\pm3 \\ \end{array}$	$\begin{array}{c} 6.9 \pm 1.6 \\ 6.4 \pm 1.3 \\ 6.6 \pm 2.2 \\ 6.4 \pm 1.4 \\ 3.5 \pm 2.0 \\ 1.4 \pm 0.5 \\ 3.1 \pm 0.7 \\ 5.3 \pm 1.9 \\ - \\ \hline \mathbf{Th} \\ 0.98 \pm 0.03 \\ 0.59 \pm 0.02 \\ 0.74 \pm 0.04 \\ 0.54 \pm 0.03 \\ 0.68 \pm 0.04 \\ 0.59 \pm 0.04 \\ \end{array}$	$\begin{array}{c} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ 10\pm0.8\\ 24\pm3\\ 18\pm2\\ \hline {\bf Ti}^*\\ 129\pm14\\ 115\pm32\\ 100\pm3\\ 86\pm5\\ 72\pm4\\ 61\pm3\\ \end{array}$	$\begin{array}{c} \textbf{Sb} \\ 0.71 \pm 0.05 \\ 0.48 \pm 0.03 \\ 0.25 \pm 0.01 \\ 0.32 \pm 0.02 \\ 0.13 \pm 0.01 \\ 0.31 \pm 0.02 \\ 0.16 \pm 0.01 \\ 0.45 \pm 0.03 \\ 1.2 \pm 0.05 \\ 0.91 \pm 0.08 \\ \hline \textbf{V*} \\ 3.6 \pm 0.6 \\ 2.9 \pm 0.8 \\ 3.1 \pm 0.3 \\ 2.7 \pm 0.2 \\ 2.0 \pm 0.3 \\ 1.7 \pm 0.3 \end{array}$	$\begin{array}{c} 0.86\pm 0.03 \\ 0.62\pm 0.03 \\ 0.55\pm 0.03 \\ 0.55\pm 0.02 \\ 0.45\pm 0.02 \\ 0.45\pm 0.02 \\ 0.49\pm 0.02 \\ 0.58\pm 0.03 \\ 0.74\pm 0.03 \\ 1.6\pm 0.08 \\ 1.8\pm 0.03 \\ \hline \textbf{Zn} \\ 55\pm 2 \\ 56\pm 2 \\ 38\pm 2 \\ 51\pm 3 \\ 46\pm 3 \\ 56\pm 2 \\ \hline \end{array}$
$ \begin{array}{c} 1\\ 2\\ 3\\ 4\\ 5\\ 6\\ 7\\ 8\\ 9\\ 10\\ 1\\ 2\\ 3\\ 4\\ 5\\ 6\\ 7\\ 6\\ 6\\ 7\\ 6\\ 6\\ 7\\ 6\\ 6\\ 6\\ 7\\ 6\\ 6\\ 6\\ 7\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\$	Mn* 520±21 509±52 415±10 448±2 290±6 390±33 504±20 273±20 - - Se* 4.5±1.3 5.3±0.7 4.8±0.7 5.2±1.1 4.3±1.0 4.0±1.0 4.5±1.0	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 1.9{\pm}0.1\\ 72{\pm}7.0\\ 4.1{\pm}0.5\\ \hline {\bf Si^*}\\ 1271{\pm}85\\ 1154{\pm}178\\ 1166{\pm}100\\ 1046{\pm}144\\ 803{\pm}138\\ 643{\pm}51\\ 925{\pm}113\\ \end{array}$	$\begin{array}{c} 1174{\pm}36\\ 832{\pm}53\\ 229{\pm}14\\ 785{\pm}55\\ 1104{\pm}75\\ 474{\pm}29\\ 706{\pm}76\\ 996{\pm}46\\ 3075{\pm}254\\ 2476{\pm}156\\ \textbf{Sm}\\ 1.3{\pm}0.05\\ 0.55{\pm}0.02\\ 0.62{\pm}0.06\\ 0.39{\pm}0.03\\ 0.64{\pm}0.045\\ 0.51{\pm}0.029\\ 0.43{\pm}0.038\\ \end{array}$	$\begin{array}{r} Ni^{*} \\ 13\pm0.7 \\ 10\pm0.6 \\ 4.8\pm0.3 \\ 10\pm0.9 \\ 4.8\pm0.2 \\ 3.9\pm0.4 \\ 9.8\pm0.8 \\ 7.8\pm0.4 \\ 59\pm3 \\ 85\pm5 \\ \hline Sr \\ 68\pm5 \\ 59\pm3 \\ 34\pm2 \\ 46\pm4 \\ 24\pm2 \\ 29\pm3 \\ 49\pm4 \\ \end{array}$	$\begin{array}{c} 6.9 \pm 1.6 \\ 6.4 \pm 1.3 \\ 6.6 \pm 2.2 \\ 6.4 \pm 1.4 \\ 3.5 \pm 2.0 \\ 1.4 \pm 0.5 \\ 3.1 \pm 0.7 \\ 5.3 \pm 1.9 \\ - \\ \hline \mathbf{Th} \\ 0.98 \pm 0.03 \\ 0.59 \pm 0.02 \\ 0.74 \pm 0.04 \\ 0.54 \pm 0.03 \\ 0.68 \pm 0.04 \\ 0.59 \pm 0.04 \\ 0.65 \pm 0.05 \end{array}$	$\begin{array}{c} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ 10\pm0.8\\ 24\pm3\\ 18\pm2\\ \hline {\bf Ti}*\\ 129\pm14\\ 115\pm32\\ 100\pm3\\ 86\pm5\\ 72\pm4\\ 61\pm3\\ 73\pm6\\ \end{array}$	$\begin{array}{c} \textbf{Sb} \\ 0.71 \pm 0.05 \\ 0.48 \pm 0.03 \\ 0.25 \pm 0.01 \\ 0.32 \pm 0.02 \\ 0.13 \pm 0.01 \\ 0.31 \pm 0.02 \\ 0.16 \pm 0.01 \\ 0.45 \pm 0.03 \\ 1.2 \pm 0.05 \\ 0.91 \pm 0.08 \\ \hline \textbf{V*} \\ 3.6 \pm 0.6 \\ 2.9 \pm 0.8 \\ 3.1 \pm 0.3 \\ 2.7 \pm 0.2 \\ 2.0 \pm 0.3 \\ 1.7 \pm 0.3 \\ 1.7 \pm 0.1 \\ \end{array}$	$\begin{array}{c} 0.86{\pm}0.03\\ 0.62{\pm}0.03\\ 0.55{\pm}0.03\\ 0.55{\pm}0.02\\ 0.45{\pm}0.02\\ 0.45{\pm}0.02\\ 0.58{\pm}0.03\\ 0.74{\pm}0.03\\ 1.6{\pm}0.08\\ 1.8{\pm}0.03\\ \hline {\bf Zn}\\ 55{\pm}2\\ 56{\pm}2\\ 38{\pm}2\\ 51{\pm}3\\ 46{\pm}3\\ 56{\pm}2\\ 126{\pm}3\\ \end{array}$
$ \begin{array}{c} 1\\ 2\\ 3\\ 4\\ 5\\ 6\\ 7\\ 8\\ 9\\ 10\\ 1\\ 2\\ 3\\ 4\\ 5\\ 6\\ 7\\ 8\\ 8\\ 8\\ 8\\ 8\\ 8\\ 8\\ 8\\ 8\\ 8\\ 8\\ 8\\ 8\\$	Mn* 520±21 509±52 415±10 448±2 290±6 390±33 504±20 273±20 - - Se* 4.5±1.3 5.3±0.7 4.8±0.7 5.2±1.1 4.3±1.0 4.0±1.0	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.05\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ 1.9{\pm}0.1\\ 72{\pm}7.0\\ 4.1{\pm}0.5\\ {\bf Si^*}\\ 1271{\pm}85\\ 1154{\pm}178\\ 1166{\pm}100\\ 1046{\pm}144\\ 803{\pm}138\\ 643{\pm}51\\ \end{array}$	$\begin{array}{c} 1174{\pm}36\\ 832{\pm}53\\ 229{\pm}14\\ 785{\pm}55\\ 1104{\pm}75\\ 474{\pm}29\\ 706{\pm}76\\ 996{\pm}46\\ 3075{\pm}254\\ 2476{\pm}156\\ \textbf{Sm}\\ 1.3{\pm}0.05\\ 0.55{\pm}0.02\\ 0.62{\pm}0.06\\ 0.39{\pm}0.03\\ 0.64{\pm}0.045\\ 0.51{\pm}0.029\\ 0.43{\pm}0.038\\ 0.95{\pm}0.056\\ \end{array}$	$\begin{array}{r} Ni^{*} \\ 13\pm0.7 \\ 10\pm0.6 \\ 4.8\pm0.3 \\ 10\pm0.9 \\ 4.8\pm0.2 \\ 3.9\pm0.4 \\ 9.8\pm0.8 \\ 7.8\pm0.4 \\ 59\pm3 \\ 85\pm5 \\ \hline Sr \\ 68\pm5 \\ 59\pm3 \\ 34\pm2 \\ 46\pm4 \\ 24\pm2 \\ 29\pm3 \\ 49\pm4 \\ 46\pm4 \\ \end{array}$	$\begin{array}{c} 6.9 \pm 1.6 \\ 6.4 \pm 1.3 \\ 6.6 \pm 2.2 \\ 6.4 \pm 1.4 \\ 3.5 \pm 2.0 \\ 1.4 \pm 0.5 \\ 3.1 \pm 0.7 \\ 5.3 \pm 1.9 \\ - \\ - \\ \hline \mathbf{Th} \\ 0.98 \pm 0.03 \\ 0.59 \pm 0.02 \\ 0.74 \pm 0.04 \\ 0.54 \pm 0.03 \\ 0.68 \pm 0.04 \\ 0.59 \pm 0.04 \\ 0.65 \pm 0.05 \\ 0.95 \pm 0.05 \\ \end{array}$	$\begin{array}{c} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ 10\pm0.8\\ 24\pm3\\ 18\pm2\\ \hline {\bf Ti}^*\\ 129\pm14\\ 115\pm32\\ 100\pm3\\ 86\pm5\\ 72\pm4\\ 61\pm3\\ \end{array}$	$\begin{array}{c} \textbf{Sb} \\ 0.71\pm 0.05 \\ 0.48\pm 0.03 \\ 0.25\pm 0.01 \\ 0.32\pm 0.02 \\ 0.13\pm 0.01 \\ 0.31\pm 0.02 \\ 0.16\pm 0.01 \\ 0.45\pm 0.03 \\ 1.2\pm 0.05 \\ 0.91\pm 0.08 \\ \hline \textbf{V*} \\ 3.6\pm 0.6 \\ 2.9\pm 0.8 \\ 3.1\pm 0.3 \\ 2.7\pm 0.2 \\ 2.0\pm 0.3 \\ 1.7\pm 0.1 \\ 3.5\pm 0.3 \\ \end{array}$	$\begin{array}{c} 0.86{\pm}0.03\\ 0.62{\pm}0.03\\ 0.55{\pm}0.03\\ 0.55{\pm}0.02\\ 0.45{\pm}0.02\\ 0.45{\pm}0.02\\ 0.58{\pm}0.03\\ 0.74{\pm}0.03\\ 1.6{\pm}0.08\\ 1.8{\pm}0.03\\ \hline \textbf{Zn}\\ 55{\pm}2\\ 56{\pm}2\\ 38{\pm}2\\ 51{\pm}3\\ 46{\pm}3\\ 56{\pm}2\\ 126{\pm}3\\ 67{\pm}4\\ \end{array}$
$ \begin{array}{c} 1\\ 2\\ 3\\ 4\\ 5\\ 6\\ 7\\ 8\\ 9\\ 10\\ 1\\ 2\\ 3\\ 4\\ 5\\ 6\\ 7\\ 6\\ 6\\ 7\\ 6\\ 6\\ 7\\ 6\\ 6\\ 6\\ 7\\ 6\\ 6\\ 6\\ 7\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\ 6\\$	Mn* 520±21 509±52 415±10 448±2 290±6 390±33 504±20 273±20 - - Se* 4.5±1.3 5.3±0.7 4.8±0.7 5.2±1.1 4.3±1.0 4.0±1.0 4.5±1.0 4.2±1.4 -	$\begin{array}{c} 1.6{\pm}0.2\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 0.45{\pm}0.03\\ 1.7{\pm}0.1\\ 1.0{\pm}0.1\\ 1.0{\pm}0.1\\ 1.9{\pm}0.1\\ 72{\pm}7.0\\ 4.1{\pm}0.5\\ \hline {\bf Si^*}\\ 1271{\pm}85\\ 1154{\pm}178\\ 1166{\pm}100\\ 1046{\pm}144\\ 803{\pm}138\\ 643{\pm}51\\ 925{\pm}113\\ \end{array}$	$\begin{array}{c} 1174{\pm}36\\ 832{\pm}53\\ 229{\pm}14\\ 785{\pm}55\\ 1104{\pm}75\\ 474{\pm}29\\ 706{\pm}76\\ 996{\pm}46\\ 3075{\pm}254\\ 2476{\pm}156\\ \textbf{Sm}\\ 1.3{\pm}0.05\\ 0.55{\pm}0.02\\ 0.62{\pm}0.06\\ 0.39{\pm}0.03\\ 0.64{\pm}0.045\\ 0.51{\pm}0.029\\ 0.43{\pm}0.038\\ \end{array}$	$\begin{array}{r} Ni^{*} \\ 13\pm0.7 \\ 10\pm0.6 \\ 4.8\pm0.3 \\ 10\pm0.9 \\ 4.8\pm0.2 \\ 3.9\pm0.4 \\ 9.8\pm0.8 \\ 7.8\pm0.4 \\ 59\pm3 \\ 85\pm5 \\ \hline Sr \\ 68\pm5 \\ 59\pm3 \\ 34\pm2 \\ 46\pm4 \\ 24\pm2 \\ 29\pm3 \\ 49\pm4 \\ \end{array}$	$\begin{array}{c} 6.9 \pm 1.6 \\ 6.4 \pm 1.3 \\ 6.6 \pm 2.2 \\ 6.4 \pm 1.4 \\ 3.5 \pm 2.0 \\ 1.4 \pm 0.5 \\ 3.1 \pm 0.7 \\ 5.3 \pm 1.9 \\ - \\ \hline \mathbf{Th} \\ 0.98 \pm 0.03 \\ 0.59 \pm 0.02 \\ 0.74 \pm 0.04 \\ 0.54 \pm 0.03 \\ 0.68 \pm 0.04 \\ 0.59 \pm 0.04 \\ 0.65 \pm 0.05 \end{array}$	$\begin{array}{c} 18\pm0.9\\ 20\pm1\\ 13\pm0.8\\ 17\pm1.3\\ 6.8\pm0.4\\ 10\pm1.1\\ 15\pm1.2\\ 10\pm0.8\\ 24\pm3\\ 18\pm2\\ \hline {\bf Ti}*\\ 129\pm14\\ 115\pm32\\ 100\pm3\\ 86\pm5\\ 72\pm4\\ 61\pm3\\ 73\pm6\\ \end{array}$	$\begin{array}{c} \textbf{Sb} \\ 0.71 \pm 0.05 \\ 0.48 \pm 0.03 \\ 0.25 \pm 0.01 \\ 0.32 \pm 0.02 \\ 0.13 \pm 0.01 \\ 0.31 \pm 0.02 \\ 0.16 \pm 0.01 \\ 0.45 \pm 0.03 \\ 1.2 \pm 0.05 \\ 0.91 \pm 0.08 \\ \hline \textbf{V*} \\ 3.6 \pm 0.6 \\ 2.9 \pm 0.8 \\ 3.1 \pm 0.3 \\ 2.7 \pm 0.2 \\ 2.0 \pm 0.3 \\ 1.7 \pm 0.3 \\ 1.7 \pm 0.1 \\ \end{array}$	$\begin{array}{c} 0.86{\pm}0.03\\ \hline 0.62{\pm}0.03\\ \hline 0.62{\pm}0.03\\ \hline 0.55{\pm}0.03\\ \hline 0.55{\pm}0.02\\ \hline 0.45{\pm}0.02\\ \hline 0.49{\pm}0.02\\ \hline 0.58{\pm}0.03\\ \hline 0.74{\pm}0.03\\ \hline 1.6{\pm}0.08\\ \hline 1.8{\pm}0.03\\ \hline \textbf{Zn}\\ \hline 55{\pm}2\\ \hline 56{\pm}2\\ \hline 38{\pm}2\\ \hline 51{\pm}3\\ \hline 46{\pm}3\\ \hline 56{\pm}2\\ \hline 126{\pm}3\\ \end{array}$

* Determined by AES

In comparison with heavy metal concentrations in some European countries (table 4), it is evident that concentration values of some heavy metals are about equal. Only the concentration of Cr was determined higher in Russia than in Europe.

Table 4. Heavy metal concentrations (maximum values) mg/kg in Europe 2005/2006 (Harmens et. al. 2008)

Country	Element				
	Pb	Си	Cr	Zn	Ni
Lithuania	7.67	11.6	2.28	36.7	1.98
Latvia	50.0	12.0	5.0	280.0	5.34
Estonia	3.25	7.05	1.47	24.3	1.83

However, it should be stressed that surveys in Europe have been conducted with different species of mosses and with specific methodology (Harmens et. al. 2010).

4. Conclusions

- 1. To monitor the atmosphere pollution, one should use one species of moss because different species of mosses (even those belonging to the same systematic groups) have differences in accumulation capacity.
- 2. Epiphyte moss *Pylaisia polyantha* is the most wide-spread species and has highest accumulating capacity. Therefore, this moss is regarded as the optimal test-object to monitor the atmospheric pollution. The use of epiphytic mosses enables monitoring of both forest and urbanized areas.
- 3. It is recommended to use moss species *Dicranum polysetum* having the highest accumulation capacity for regional monitoring of atmosphere (when terrestrial mosses are used).

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Nadezhda K. Ryzhakova - candidate of physical and mathematical sciences, docent in National Research Tomsk Polytechnic University, Department of Applied Physics Engineering.

Address:	Institute of Physics and Technology,
	Tomsk Polytechnic University,
	Tomsk, RF
E-mail:	nkr@tpu.ru

Natalya S. Rogova - candidate of engineering sciences, assistant, in National Research Tomsk Polytechnic University, Department of Applied Physics Engineering.

Address:	Institute of Physics and Technology,		
	Tomsk	Polytechnic	University,
	Tomsk, I	RF	
E-mail:	rogova@	interact.phtd.t	pu.ru

Alex L. Borisenko - candidate of biological sciences, docent, in National Research Tomsk State University, Institute of Biology.

Address:	Institute of Biology, Tomsk State
	University, Tomsk, RF
E-mail:	alb@sibmail.com

Samanų akumuliacinių savybių, naudojamų regioniniams ir vietiniams atmosferos teršalams įvertinti, tyrimas

Nadezhda K. Ryzhakova¹, Natalia S. Rogova¹, Alex L. Borisenko²

¹Fizikos ir technologijų institutas, Tomsko politechnikos universitetas, Tomskas, Rusija ²Biologijos institutas, Tomsko valstybinis universitetas, Tomskas, Rusija

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Atmosferos sunkiujų metalų ir kitų toksinių elementų nuosėdų monitoringas, naudojant samanas kaip bioindikatorių, yra plačiai naudojamas visame pasaulyje. Tyrimo tikslas – nustatyti tinkamiausias samanų rūšis pagal jų akumuliacines savybes. Šioje studijoje buvo atliktas sunkiųjų metalų ir kitų toksinių elementų akumuliacijos plačiai paplitusiose viduržemio samanose tyrimas. Tyrimo pagrindinis tikslas buvo įvertinti 4 skirtingų sausumos samanų rūšių, 4 vandeningų vietovių samanų rūšių bei 2 epifitinių samanų rūšių akumuliacines savybes. Naudojant neutronų aktyvavimo analizę (NAA) ir atomo emisijos spektrometriją (AES) samanų meginiuose buvo nustatytos 32 elementų koncentracijos. Tyrimas parodė, kad akumuliacinės savybės gali atskleisti reikšmingus skirtumus lyginant tarprūšines ir rūšines savybes. Dicranum polysetum akumuliacinis pajėgumas buvo didesnis nei kitų sausumos samanų rūšių, taip pat kaip Aulacomnium palustre didesnis nei kitų vandeningų vietovių samanų. Ši samanos savybė buvo panaudota atmosferiniam teršalų monitoringui didžiulėje teritorijoje ypač nustatyti tarpvalstybinę sunkiųjų metalų taršos pernašą. Akumuliacinė epifitinių samanų savybė buvo didesnė nei visų kitų. Epifitinės samanos aptinkamos ant senų medžių žievės (drebulių, tuopų, beržų), dažnai pastebimos miestų teritorijose. Būtent dėl šios priežasties epifitinės samanos gali būti naudojamos ne vien atmosferos teršalų stebėsenai didelėse teritorijose, tačiau ir tiriant atmosferos taršą pramoniniuose rajonuose, taip pat gyvenamosiose vietovėse, norint įvertinti atmosferos užterštumą dėl vietinio taršos šaltinio.